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Test B, continued 18.

The pressure exerted by each gas in a mixture is called the of that gas. 19. If the

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Answers
temperature and
number of moles of a
gas remain constant
but the volume
increases, the pressure
of the gas will . 20. The
lowest possible
temperature,
corresponding to zero
on the kelvin scale, is

**Assessment Chapter
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Holt McDougal Modern
Chemistry Chapter Test
Assessment Chapter
Test B Teacher Notes

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and Answers 5 The

Periodic Law TEST B 1.
a 2. c 3. d 4. d 5. a 6. a
7. c 8. a 9. lanthanides
10. 2 11. fourth 12.
transition elements 13.
32 14. valence
electrons 15. electron
affinity 16.
electronegativity 17.
ionization energy 18. 3
s² 3p⁴ 19. atomic
radius 20 ...

**Assessment Chapter
Test B - Wag & Paws**

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Chemistry Chapter Test
B Measurements And
Calculations Answer
Key Dalton's theory
agreed with the
modern atomic theory
in almost all cases. List
the two statements
that were later found
to be in error. All atoms
of the same element
have the same mass;
atoms cannot be
subdivided.

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Chapter Test B

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**Measurements And
Answers**

Chapter Test B,
continued 15. The
energy state of an
atom is called its
ground state. 16.

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Chapter Test B,
continued 28. Write the
general equilibrium
expression for the
dissociation of an acid-
base indicator that is a

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Answers
weak acid, HIn, and explain how this equilibrium determines the color of the indicator at a given pH.
PART IV In the space provided, identify each of the following substances as acidic, basic, or

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MARSHALL'S
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Test B, continued 7.

The discovery of the nucleus was a result of Rutherford's observation that a small percentage of the positively charged particles bombarding the metal's surface a. were slightly deflected as they passed through the metal.

**Chapter Test B -
PC\|MAC**

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Chapter Test Name
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Test B, continued 17. A
substance combines
with oxygen, releasing
a large amount of
energy as heat and
light, in a(n) . 18. The
decomposition of a
substance by an
electric current is
called . 19. A(n) orders
the elements by the
ease with which they
undergo certain
chemical reactions. 20.

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**Assessment Chapter
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Holt McDougal Modern
Chemistry 3 Chapter
Test Chapter Test B,
continued 16 Modern
chemistry chapter 3
test b answers. The
measure of the ability
of an atom in a
chemical compound to
attract electrons from
another atom in the
compound is called
_____. 17. The energy
required to remove one

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electron from an atom
is called its _____. 18.

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Chapter 3 Test B
Answers

Particle Mass number
19. Proton 20. Neutron
21. Electron 10

CHAPTER 3 TEST

Relative charge

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Class CHAPTER 3 TEST
continued SHORT
ANSWER Write the
answers to the
following questions in
the space provided. 22.

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Test B, continued 17. A

substance combines
with oxygen, releasing

a large amount of

energy as heat and

light, in a (n). 18. The

decomposition of a

substance by an

electric current is

called.

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CHAPTER 3 TEST

continued Date Class

FILL IN THE BLANK

Write the correct term (or terms) in the space provided. 9, If a particular compound is composed of elements A and B, the ratio of the mass of B to the mass of A will always be the same. This is a statement of the law of exactly 12 g of carbon-12 is referred to as a(n) 11.

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Chapter Test Name

_____ Class _____ Date
_____ Chapter Test A,

continued Use this
figure to answer
questions 7 and 8.

_____ 7. A solution
containing 35 g of
 Li_2SO_4 dissolved in
100 g of water is
heated from 10°C
to 90°C .

According to
information in the

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figure, this
temperature change
would result in a. an
additional ...

**Chapter 12 Practice
Test - Doral
Academy
Preparatory**

Test. PLAY. Match.
Gravity. Created by.
jkinsler12. Gases and
Gas Laws. Terms in this
set (26) Ideal Gas Law.
... temperature and
pressure are inversely
proportional to the

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square roots of their
molar masses Rate of
effusion of A / B = $\sqrt{M_b}$
/ $\sqrt{M_a}$. Kinetic-
Molecular Theory ...
Chapter 10: Modern
Chemistry vocab. 45
terms. emitch4243.
Modern ...

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Chapter 11
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CHAPTER 22 TEST
Nuclear Chemistry
Class MULTIPLE CHOICE

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Answers

On the line at the left of each statement, write the letter of the choice that best completes the statement or answers the question. After converting units, the nuclear mass defect is equivalent to the a. atomic mass b. electrostatic force c. energy of chemical reaction

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